

## Aurintricarboxylic Acid-Derived Polysalicylates as Platforms for Drug Development: A Mini-Review

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**Abstract:** Aurintricarboxylic acid, a dye developed for the colorimetric assay of aluminum ion, has diverse effects on many biological systems. Many of these effects are due to the complex chemical composition of the commercial grade material used in these studies. These components include small molecular weight species as well as progressively larger complexes derived from salicylate residues. Since certain pharmacological properties have been linked to structural subsets of the complex commercial mixture, the aurintricarboxylic acid-derived polysalicylates offer a unique opportunity for drug development.

**Keywords:** aurintricarboxylic acid, polysalicylates, polymers, drug development

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### I. Introduction

#### Chemistry of Aurintricarboxylic Acid-Derived Polysalicylates

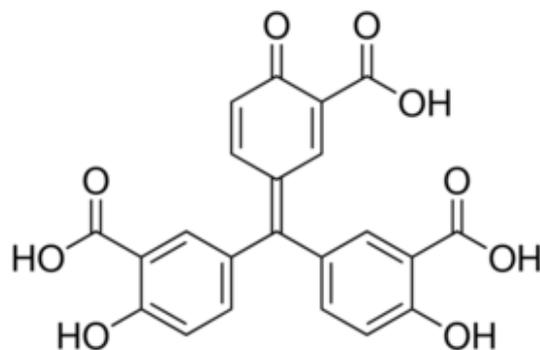
Aurintricarboxylic acid (ATA) was originally synthesized for the assay of aluminum ion. While variations exist in the preparation of ATA, it is derived from the reaction of salicylic acid with sodium nitrite and sulfuric acid<sup>1</sup>. It is more commonly available as an ammonium salt, although free acid and sodium salts have been commercially available. While commercial preparations of ATA contain the triphenylmethane dye (Figure no 1), the preparations used by most investigators for testing ATA for biochemical activity over several decades contain other components; including polymeric forms that may be generally described as polysalicylates, due to the covalent aggregation of salicylate moieties. An example of one of these polysalicylates is shown in Figure no 2<sup>2</sup>. These aggregates may form during synthesis and freely in various aqueous systems<sup>2</sup>.

### II. Selected Activities of ATA

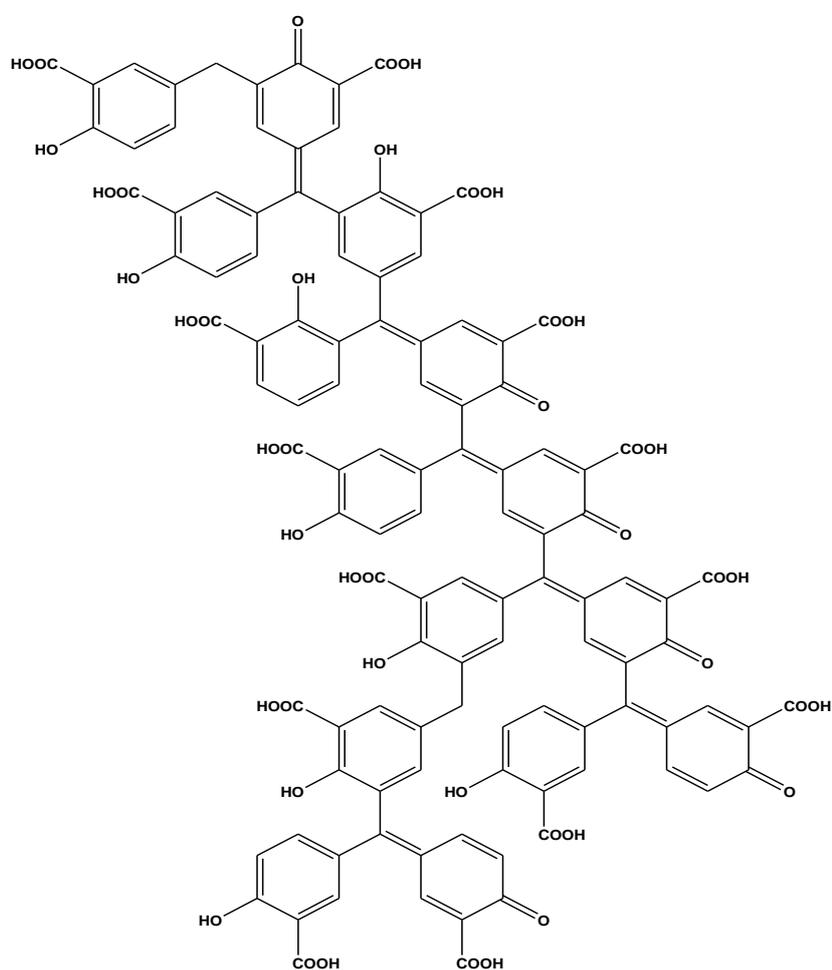
#### *In Vitro and In Vivo*

In retrospect, the complex composition of commercial ATA contributed to the wide array of biochemical activities found through *in vitro* studies (Table no 1)<sup>2,4-13</sup>. These biochemical activities include, but are not limited to inhibition of nuclease and protein synthesis as well as anti-apoptotic effects<sup>4,8,10</sup>. These may play a role in the *in vitro* actions seen with ATA, which include antiviral<sup>2,5,6,7</sup>, antiparasitic<sup>9</sup>, antitumor<sup>13</sup> and antimicrobial activities<sup>11,12</sup>. While many investigators note that the commercial preparations of ATA they use are complex mixtures, it is clear that isolation of the active agent responsible for the activity they observe may be more formidable than characterization of ATA's observed effects. In addition to the *in vitro* activities, the ability of commercial ATA to produce effects *in vivo* is promising (Table no 2)<sup>9,14-19</sup>. These include protection of neuronal ischemia<sup>14</sup>, endotoxin shock<sup>15</sup> and beryllium poisoning<sup>16</sup>. Antiplatelet<sup>17,18,19</sup> and anti-Cryptosporidium activities<sup>9</sup> have been characterized *in-vivo*. Attempts to isolate components responsible for a few of these activities have been restricted to a much smaller group of laboratories, and rarely to homogeneity<sup>2,3,19</sup>.

**Figure no 1:** Aurintricarboxylic Acid.



**Figure no 2:** An aurintricarboxylic acid polysalicylate (adapted from Cushman, et al.,<sup>2)</sup>



**Table no 1:** Examples of ATA activities *in vitro*

ACTIVITY	REFERENCE
Inhibition of Nuclease	4
Antiviral	
Influenza	5
Enterovirus	6
Vaccinia	7
HIV	2
Inhibition of Apoptosis	8
Anti-Cryptosporidium	9
Inhibition of Protein Synthesis	10
Reduction in Microbial Virulence	
Staphylococcus sp.	11
Yersinia pestis	12
Antitumor	13

**Table no 2:** Examples of ATA activities *in vivo*.

ACTIVITY	REFERENCE
Anti-Cryptosporidium	9
Protection from Neuronal Ischemia	14
Protection from Endotoxin Shock	15
Protection from Beryllium Poisoning	16
Anti-Platelet	17,18,19

### III. Opportunities and Challenges

It is remarkable that a polymeric variation in salicylate moieties can result in such an extensive pharmacological profile. While the complex mixture of ATA is suitable for its application as a complexation reagent, this is problematic for drug development. Isolation and characterization of a single molecular structure associated with a pharmacological activity is often required for successful manufacturing, quality control, drug approval and clinical applications. In addition, the reactive nature of ATA in solution must be considered during drug development processes, even when a pure form of a component is isolated initially. The challenge associated with the diverse chemical composition of ATA in commercial preparations extends to potential drug delivery systems, when considering the activity and polyanionic nature of ATA<sup>20,21,22</sup>. Isolation of ATA components associated with the aforementioned activities, while daunting, offers an unusual opportunity to model the interaction of various ligands with molecular targets as templates for drug design. While the components of ATA may not be ideal as drug candidates, their derivatives and analogues identified through molecular modeling studies may be important alternatives. Clearly ATA components remain as valuable molecular probes in drug development.

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